

Efficient and Copper-Free Sonogashira Cross-Coupling Reaction Catalyzed by Pd(OAc)₂/Pyrimidines Catalytic System

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Keywords: Palladium / Cross-coupling reaction / Homogeneous catalysis

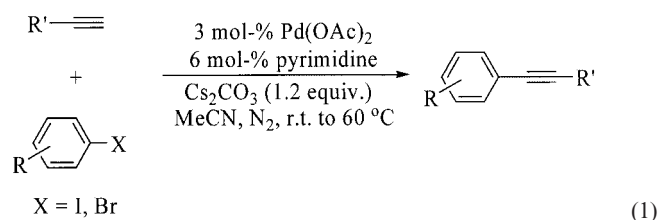
An efficient and copper-free catalyst system for Sonogashira cross-coupling reaction is presented. In the presence of 3 mol-% of Pd(OAc)₂ and 6 mol-% of 2-aminopyrimidine-4,6-diol, various aryl iodides and bromides can be coupled smoothly with terminal alkynes to afford the corresponding

coupled alkynes in moderate to excellent yields under mild conditions.

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Introduction

The acetylenic subunit is a commonly found motif in biologically active molecules and a useful building block in organic synthesis. Accordingly, the construction of alkynes remains an active area in organic synthesis. Besides the homocoupling reactions of terminal alkynes, the Pd-catalyzed Sonogashira cross-coupling reaction represents another important method for the formation of alkynes.^[1] Generally, the combination of palladium, phosphane and CuI is used as the catalytic system for the reaction.^[2–5] However, many phosphane ligands are air-sensitive and expensive.^[6] Besides these drawbacks, the other significant limits on their synthetic applications is that the generation of some Cu^I acetylides in situ by the reaction of CuI with alkyne can result in oxidative homocoupling reaction of alkynes readily.^[5f,7] For these reasons, the development of an efficient, inexpensive, copper- and phosphane-free catalytic system is still significant. Generally, a number of *N*-heterocyclic compounds is commercially available and these compounds (often pyridines) can directly be used as ligands for palladium catalysts.^[2c] Thus, we tested these ligands in palladium-catalyzed cross-coupling reactions. Indeed, we found that pyrimidines were efficient ligands for Pd(OAc)₂-catalyzed Sonogashira cross-coupling reaction of aryl halides with terminal alkynes. Here we report our initial results according to Equation (1).



Results and Discussion

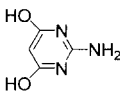
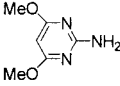
Efficiency of Pd(OAc)₂/pyrimidines as the catalytic system for coupling of 1-iodo-4-methoxybenzene (**1a**) with phenylacetylene (**2a**) was first evaluated, and the results are summarized in Table 1. Initially, a series of pyrimidines (**L**₁–**L**₄) as ligands were tested (entries 1–5). The results showed that the ligand 2-aminopyrimidine-4,6-diol (**L**₁) was most effective, and the amount of **L**₁ also affected the yield to some extent. Without any ligands, treatment of **1a** with **2a**, Pd(OAc)₂ and Cs₂CO₃ afforded a low yield of the corresponding cross-coupled product **3** at room temperature (entry 1), whereas the yield of **3** was increased sharply to 80% when 3 mol-% of **L**₁ was added (entry 2). In the presence of 3 mol-% of Pd(OAc)₂ and 6 mol-% of **L**₁, the yield of **3** was enhanced to 92% (entry 3). Identical results were observed when the amount of **L**₁ was further increased to 12 mol-% (entry 4). A set of bases, such as Cs₂CO₃, KOAc, K₂CO₃, and Et₃N, were then investigated, and Cs₂CO₃ as the base gave the best results (entries 3 and 8–10). Finally, some solvents including MeCN, acetone and DMF were also examined and MeCN as the solvent provided the highest yield (entries 3, 11, and 12).

As shown in Table 2, the combination of Pd(OAc)₂ and 2-aminopyrimidine-4,6-diol (**L**₁) is an effective catalytic system for the cross-coupling reactions of various aryl halides **1a–j** with alkynes **2a–e**. In the presence of 3 mol-% of Pd(OAc)₂, 6 mol-% of **L**₁, and 1.2 equiv. of Cs₂CO₃, coupling of a

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Table 1. Palladium-catalyzed Sonogashira cross-coupling reaction of 1-iodo-4-methoxybenzene (**1a**) with phenylacetylene (**2a**).^[a]

Entry	Ligand	Base	Time [h]	Yield [%] ^[b]
1	—	Cs ₂ CO ₃	15	38
2 ^[c]	 (L ₁)	Cs ₂ CO ₃	8	80
3	L ₁	Cs ₂ CO ₃	4	92
4 ^[d]	L ₁	Cs ₂ CO ₃	4	91
5	 (L ₂)	Cs ₂ CO ₃	8	78
6	 (L ₃)	Cs ₂ CO ₃	8	72
7	 (L ₄)	Cs ₂ CO ₃	8	41
8	L ₁	K ₂ CO ₃	8	65
9	L ₁	NaOAc	10	40
10	L ₁	Et ₃ N	9	50
11 ^[e]	L ₁	Cs ₂ CO ₃	12	80
12 ^[f]	L ₁	Cs ₂ CO ₃	6	74

[a] Unless otherwise indicated, the reaction conditions were as follows: **1a** (1 mmol), **2a** (1.2 mmol), Pd(OAc)₂ (3.0 mol-%), ligand (6.0 mol-%), base (1.2 equiv.), and MeCN (4 mL) under N₂ at room temperature. [b] Isolated yield. [c] Ligand (3 mol-%). [d] Ligand (12 mol-%). [e] DMF (4 mL) instead of MeCN. [f] Acetone (4 mL) instead of MeCN.

number of aryl iodides **1a–f** with alkynes **2a–c** efficiently yields the corresponding alkynes **4–11** in good yields (entries 1–8). For example, when 1-iodo-4-nitrobenzene (**1b**) is coupled with alkynes **2a–c** the corresponding products **5–7** are obtained in yields of 100%, 95%, and 82%, respectively (entries 2–4). The results indicate that the efficiency of the Pd(OAc)₂/**L**₁ catalytic system for the reactions of aryl bromides **1g–j** is decreased to some extent, and both higher reaction temperature and prolonged reaction time are required (entries 9–16). For example, treatment of 1-bromo-4-nitrobenzene (**1g**) with alkynes **1a**, **1b**, **1d**, or **1e** gives moderate to good yields of the corresponding products after 20 h at 60 °C (entries 9–12), whereas the deactivated iodide **1f** is consumed completely within 9 h at room temperature (82% yield, entry 8). Unfortunately, the coupling of aryl chlorides **1k** and **1l** with **2a**, respectively, was unsuccessful (entries 17 and 18).

Conclusions

In summary, we have developed an efficient Pd(OAc)₂/2-aminopyrimidine-4,6-diol catalytic system for the Sonogashira cross-coupling reaction. In the presence of Pd(OAc)₂

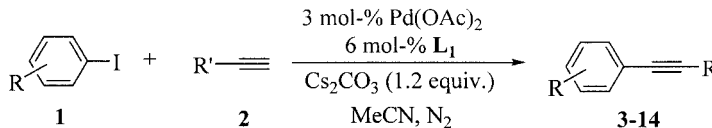
and 2-aminopyrimidine-4,6-diol, the reactions of a number of aryl iodides and bromides with terminal alkynes were carried out efficiently to afford the desired products in moderate to excellent yields under mild conditions. Further efforts to extend the application of this catalytic system and this protocol in organic synthesis are underway in our laboratory.

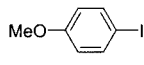
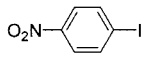
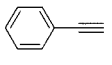
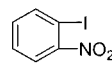
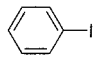
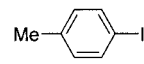
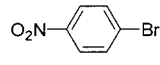
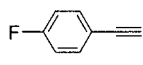
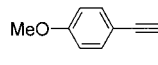
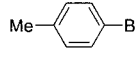
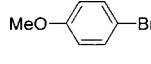
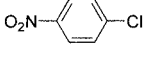
Experimental Section

General Remarks: ¹H and ¹³C NMR spectra were recorded on an INOVA-400 (Varian) spectrometers and a Bruker AMX-300 spectrometers with the use of CDCl₃ as the solvent. All reagents were directly used as obtained commercially. All products **3–14** are known,^[3–5] and their analytical data and spectra (¹H and ¹³C NMR) are available as Supporting Information (see also the footnote on the first page of this article).

Typical Experimental Procedure for the Pd(OAc)₂/L**₁-Catalyzed Sonogashira Cross-Coupling Reaction:** A mixture of aryl halide **1** (1.0 mmol), alkyne **2** (1.2 mmol), Pd(OAc)₂ (3.0 mol-%), 2-aminopyrimidine-4,6-diol (**L**₁, 6.0 mol-%), Cs₂CO₃ (1.2 equiv.), and MeCN (4 mL) was stirred under N₂ at the indicated reaction temperature for desired time until complete consumption of starting material as monitored by TLC. After the mixture was filtered and

Table 2. Palladium-catalyzed Sonogashira coupling reaction in the presence of 2-aminopyrimidine-4,6-diol (**L**₁).^[a]



Entry	ArX	Alkyne	Time [h]	Yield [%] ^[b]	
1	 (1a)	<i>n</i> -C ₈ H ₁₇ -C≡CH ((2b))	7	87	4
2	 (1b)	 (2a)	5	100	5
3	1b	(2b)	5	95	6
4	1b	≡CH ₂ OH ((2c))	9	82	7
5	 (1c)	2a	6	95	8
6	 (1d)	2a	6	98	9
7	 (1e)	2a	6	92	10
8	 (1f)	2a	9	82	11
9 ^[c]	 (1g)	2a	20	95	5
10 ^[c]	1g	2b	21	65	6
11 ^[c]	1g	 (2d)	21	88	12
12 ^[c]	1g	 (2e)	24	81	13
13 ^[c]	 (1h)	2a	18	80	9
14 ^[c]	 (1i)	2a	24	60	10
15 ^[c]	(1i)	2b	24	63	14
16 ^[c]	 (1j)	2a	21	72	3
17 ^[c]	 (1k)	2a	24	10	5
18 ^[c]	 (1l)	2a	24	trace	9

[a] Unless otherwise indicated, the reaction conditions were as follows: **1** (1 mmol), **2** (1.2 mmol), Pd(OAc)₂ (3.0 mol-%), **L**₁ (6.0 mol-%), Cs₂CO₃ (1.2 equiv.), and MeCN (4 mL) at room temperature under N₂. [b] Isolated yield. [c] At 60 °C.

evaporated, the residue was purified by flash column chromatography (hexane/ethyl acetate) to afford the coupled products **3–14**.

Acknowledgments

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[1] a) H. G. Viehe, *Chemistry of Acetylene*, Marcel Dekker, New York, **1969**; b) F. Bohlmann, F. T. Burkhart, C. Zero, *Naturally Occurring Acetylenes*, Academic Press, London and New York, **1973**; c) F. Diederich, P. J. Stang, R. R. Tykwinski, *Acetylene Chemistry*, Wiley-VCH, Weinheim, **2005**; d) K. Sonogashira, in: *Comprehensive Organic Synthesis* (Eds.: B. M. Trost, I. Fleming), Pergamon, Oxford, U. K., **1991**, Vol. 3, p 551; e) F.

- Alonso, I. P. Beletskaya, M. Yus, *Chem. Rev.* **2004**, *104*, 3079–3160.
- [2] For reviews, see: a) F. Diederich, P. J. Stang, *Metal-Catalyzed Cross-coupling Reactions*, Wiley-VCH, Weinheim, **1998**; b) N. Miyaura, *Cross-Coupling Reaction*, Springer, Berlin, **2002**; c) E. Negishi, *Handbook of Organopalladium Chemistry for Organic Synthesis*, Wiley Interscience, New York, **2002**.
- [3] For recent representative papers on phosphane-palladium catalysts, see: a) V. P. W. Böhm, W. A. Herrmann, *Eur. J. Org. Chem.* **2000**, 3679–3682; b) M. R. Netherton, G. C. Fu, *Org. Lett.* **2001**, *3*, 4295–4298; c) D. Méry, K. Heuzé, D. Astruc, *Chem. Commun.* **2003**, 1934–1935; d) D. Gelman, S. L. Buchwald, *Angew. Chem. Int. Ed.* **2003**, *42*, 5993–5996; e) A. Elangovan, Y.-H. Wang, T.-I. Ho, *Org. Lett.* **2003**, *5*, 1841–1844; f) M. Feuerstein, F. Berthiol, H. Doucet, M. Santelli, *Synthesis* **2004**, 1281–1289 and references cited therein; g) C. Wolf, R. Lerebours, *Org. Biomol. Chem.* **2004**, *2*, 2161–2164; h) J.-C. Hierso, A. Fihri, R. Amardeil, P. Meunier, H. Doucet, M. Santelli, V. V. Ivanov, *Org. Lett.* **2004**, *6*, 3473–3476; i) K. Heuzé, D. Méry, D. Gauss, J.-C. Blais, D. Astruc, *Chem. Eur. J.* **2004**, *10*, 3936–3944; j) J. Cheng, Y. Sun, F. Wang, M. Guo, J.-H. Xu, Y. Pan, Z. Zhang, *J. Org. Chem.* **2004**, *69*, 5428–5432; k) M. Feuerstein, H. Doucet, M. Santelli, *Tetrahedron Lett.* **2004**, *45*, 8443–8446; l) G. Adjabeng, T. Brenstrum, C. S. Frampton, A. J. Robertson, J. Hillhouse, J. McNulty, A. Capretta, *J. Org. Chem.* **2004**, *69*, 5082–5086; m) A. Arques, D. Aunon, P. Molina, *Tetrahedron Lett.* **2004**, *45*, 4337–4340; n) N. E. Leadbeater, B. J. Tominack, *Tetrahedron Lett.* **2003**, *44*, 8653–8656.
- [4] For recent representative papers on phosphane-free ligand-palladium catalysts under copper-free conditions, see: a) J. A. Loch, M. Albrecht, E. Peris, J. Mata, J. W. Faller, R. H. Crabtree, *Organometallics* **2002**, *21*, 700–706; b) W. A. Herrmann, *Angew. Chem. Int. Ed.* **2002**, *41*, 1290–1309; c) R. A. Batey, M. Shen, A. J. Lough, *Org. Lett.* **2002**, *4*, 1411–1414; d) M. Eckhardt, G. C. Fu, *J. Am. Chem. Soc.* **2003**, *125*, 13642–13643; e) B. S. Yong, S. P. Nolan, *Chemtracts: Org. Chem.* **2003**, 205–221; f) S. Rau, K. Lamm, H. Goerls, J. Schoeffel, D. Walther, *J. Organomet. Chem.* **2004**, *689*, 3582–3592; g) R. A. Gossage, H. A. Jenkins, P. N. Yadav, *Tetrahedron Lett.* **2004**, *45*, 7689–7691; h) D. A. Alonso, L. Botella, C. Nájera, M. C. Pacheco, *Synthesis* **2004**, 1713–1718; i) V. V. Thakur, N. S. C. R. Kumar, A. Sudalai, *Tetrahedron Lett.* **2004**, *45*, 2915–2918; j) J.-H. Li, X.-D. Zhang, Y.-X. Xie, *Synthesis* **2005**, 804–808; k) J. Gil-Moltó, C. Nájera, *Eur. J. Org. Chem.* **2005**, 4073–4081.
- [5] For selected papers on the palladium-catalyzed Sonogashira cross-coupling reactions under ligand- and copper-free conditions, see: a) M. Alami, F. Ferri, G. Linstrumelle, *Tetrahedron Lett.* **1993**, *34*, 6403–6406; b) R. G. Heidenreich, K. Kohler, J. G. E. Krauter, J. Pietsch, *Synlett* **2002**, 1118–1122; c) S. Urgaonkar, J. G. Verkade, *J. Org. Chem.* **2004**, *69*, 5752–5755 and references cited therein; d) S. U. Son, Y. Jang, J. Park, H. B. Na, H. M. Park, H. J. Yun, J. Lee, T. Hyeon, *J. Am. Chem. Soc.* **2004**, *126*, 5026–5027; e) B. Liang, M. Dai, J. Chen, Z. Yang, *J. Org. Chem.* **2005**, *70*, 391–393; f) J.-H. Li, Y. Liang, Y.-X. Xie, *J. Org. Chem.* **2005**, *70*, 4393–4396; g) A. R. Gholap, K. Venkatesan, R. Pasricha, T. Daniel, R. J. Lahoti, K. V. Srinivasan, *J. Org. Chem.* **2005**, *70*, 4869–4872.
- [6] a) L. H. Pignolet, Ed., *Homogeneous Catalysis with Metal Phosphine Complexes*, Plenum, New York, **1983**; b) G. W. Parshall, S. Ittel, *Homogeneous Catalysis*, J. Wiley & Sons, New York, **1992**.
- [7] a) P. Siemsen, R. C. Livingston, F. Diederich, *Angew. Chem. Int. Ed.* **2000**, *39*, 2632–2657; b) J.-H. Li, Y. Liang, X.-D. Zhang, *Tetrahedron* **2005**, *61*, 1903–1907 and references cited therein.

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